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# Phosphorus in kidney disease: Culprit or bystander?

**P**HOSPHORUS IS ESSENTIAL for life. However, both low and high levels of phosphorus in the body have consequences, and its concentration in the blood is tightly regulated through dietary absorption, bone flux, and renal excretion and is influenced by calcitriol (1,25 hydroxyvitamin D<sub>3</sub>), parathyroid hormone, and fibroblast growth factor 23 (FGF23).

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Sekar et al,<sup>1</sup> in this issue of the *Journal*, provide an extensive review of the pathophysiology of phosphorus metabolism and strategies to control phosphorus levels in patients with hyperphosphatemia and end-stage kidney disease.

## ■ PHOSPHORUS OR PHOSPHATE?

*What's in a name? That which we call a rose  
By any other word would smell as sweet.*

—Shakespeare, *Romeo and Juliet*

The terms *phosphate* and *phosphorus* are often used interchangeably, though most writers still prefer *phosphate* over *phosphorus*.

The serum concentrations of phosphate and phosphorus are the same when expressed in millimoles per liter, as every mole of phosphate contains 1 mole of phosphorus, but not the same when expressed in milligrams per deciliter.<sup>2</sup> The molecular weight of phosphorus is 30.97, whereas the molecular weight of the phosphate ion (PO<sub>4</sub><sup>3-</sup>) is 94.97—more than 3 times higher. Therefore, using these terms interchangeably in this context can lead to numerical error.<sup>3</sup>

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Phosphorus, being highly reactive, does not exist by itself in nature and is typically present as phosphates in biologic systems. When describing phosphorus metabolism, the term *phosphates* should ideally be used because phosphates are the actual participants in the bodily processes. But in the clinical laboratory, all methods that measure serum phosphorus in fact measure inorganic phosphate and are expressed in terms of milligrams of phosphorus per deciliter rather than milligrams of phosphate per deciliter, and using these 2 terms interchangeably in clinical practice should not be of concern.<sup>4</sup>

## ■ THE PROBLEM

US adults typically ingest 1,200 mg of phosphorus each day, and about 60% to 70% of the ingested phosphorus is absorbed both by passive paracellular diffusion via tight junctions and by active transcellular transport via sodium-phosphate cotransport. The kidneys must excrete the same amount daily to maintain a steady state. As kidney function declines, phosphorus accumulates in the blood, leading to hyperphosphatemia.

Hyperphosphatemia is often asymptomatic, but it can cause generalized itching, red eyes, and adverse effects on the bone and parathyroid glands. Higher serum phosphorus levels have been shown to be associated with vascular calcification,<sup>5</sup> cardiovascular events, and higher all-cause mortality rates in the general population,<sup>6</sup> in patients with diabetes,<sup>7</sup> and in those with chronic kidney disease.<sup>8</sup> This association between higher serum phosphorus levels and the all-cause mortality rate led to the assumption that lowering serum phospho-

**As kidney  
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TABLE 1

**Phosphorus binders**

**Calcium-based**

Calcium carbonate  
Calcium acetate

**Noncalcium-based**

**Metal-based**

Aluminum hydroxide (only for short-term use)  
Lanthanum carbonate  
Iron (sucroferric oxyhydroxide, ferric citrate)

**Resin-based**

Sevelamer (sevelamer hydrochloride, sevelamer carbonate)

**Others**

Magnesium hydroxide, magnesium carbonate (risk of hypermagnesemia)  
Chitosan (minimal effect)

rus levels in these patients could reduce the rates of cardiovascular events and death, and to efforts to correct hyperphosphatemia.

Research into FGF23 continues, especially its role in cardiovascular complications of chronic kidney disease, as both phosphorus and FGF23 levels are elevated in chronic kidney disease and are implicated in poor clinical outcomes in these patients. However, both FGF23 and parathyroid hormone levels rise early in the course of kidney disease, long before overt hyperphosphatemia develops. Further, FGF23 rises earlier than parathyroid hormone and has been found to be an independent risk factor for cardiovascular events and death from any cause in end-stage kidney disease.<sup>9</sup>

Whether hyperphosphatemia is the culprit or merely an epiphenomenon of metabolic complications of chronic kidney disease is still unclear, as more molecules are being identified in the complex process of cardiovascular calcification.<sup>10</sup>

However, one thing is clear: vascular calcification is not just a simple precipitation of calcium and phosphorus. Instead, it is an active process that involves many regulators of mineral metabolism.<sup>10</sup> The complex nature of this process is likely one of the reasons that evidence is conflicting<sup>11</sup> about the benefits of phosphorus binders in terms of cardiovascular events or all-cause mortality in these patients.

**STRATEGIES TO CONTROL HYPERPHOSPHATEMIA**

**Reducing intake**

Dietary phosphorus restriction is the first step in controlling serum phosphorus. But reducing phosphorus intake while otherwise trying to optimize the nutritional status can be challenging.

The recommended daily protein intake is 1.0 to 1.2 g/kg. But phosphorus is typically found in foods rich in proteins, and restricting protein severely can compromise nutritional status and may be as bad as elevated phosphate levels in terms of outcomes.

Although plant-based foods contain more phosphate per gram of protein (ie, they have a higher ratio of phosphorus to protein) than animal-based foods, the bioavailability of phosphorus from plant foods is lower. Phosphorus in plant-based foods is mainly in the form of phytate. Humans cannot hydrolyze phytate because we lack the phytase enzyme; hence, the phosphorus in plant-based foods is not well absorbed. Therefore, a vegetarian diet may be preferable and beneficial in patients with chronic kidney disease. A small study in humans showed that a vegetarian diet resulted in lower serum phosphorus and FGF23 levels, but the study was limited by its small sample size.<sup>12</sup>

Patients should be advised to avoid foods that have a high phosphate content, such as processed foods, fast foods, and cola beverages, which often have phosphate-based food additives.

Further, one should be cautious about using supplements with healthy-sounding names. A case in point is “vitamin water”: 12 oz of this fruit punch-flavored beverage contains 392 mg of phosphorus,<sup>13</sup> and this alone would require 12 to 15 phosphate binder tablets to bind its phosphorus content.

In addition, many prescription drugs have significant amounts of phosphorus, and this is often unrecognized.

Sherman et al<sup>14</sup> reviewed 200 of the most commonly prescribed drugs in dialysis patients and found that 23 (11.5%) of the drug labels listed phosphorus-containing ingredients, but the actual amount of phosphorus was not listed. The phosphorus content ranged from 1.4 mg (clonidine 0.2 mg, Blue Point Laboratories, Dublin, Ireland) to 111.5 mg (paroxetine 40 mg, GlaxoSmith Kline, Philadelphia, PA).

**Vascular calcification is not just a simple precipitation of calcium and phosphorus**

The phosphorus content was inconsistent and varied with the dose of the agent, type of formulation (tablet or syrup), branded or generic formulation, and manufacturer.

Branded lisinopril (Merck, Kenilworth, NJ) had 21.4 mg of phosphorus per 10-mg dose, while a generic product (Blue Point Laboratories, Dublin, Ireland) had 32.6 mg. Different brands of generic amlodipine 10 mg varied in their phosphorus content from 8.6 mg (Lupin Pharmaceuticals, Mumbai, India) to 27.8 mg (Greenstone LLC, Peapack, NJ) to 40.1 mg (Qualitest Pharmaceuticals, Huntsville, AL). Rena-Vite (Cypress Pharmaceuticals, Madison, MS), a multivitamin marketed to patients with kidney disease, had 37.7 mg of phosphorus per tablet. Thus, just to bind the phosphorus content of these 3 tablets (lisinopril, amlodipine, and Rena-Vite), a patient could need at least 3 to 4 extra doses of phosphate binder.

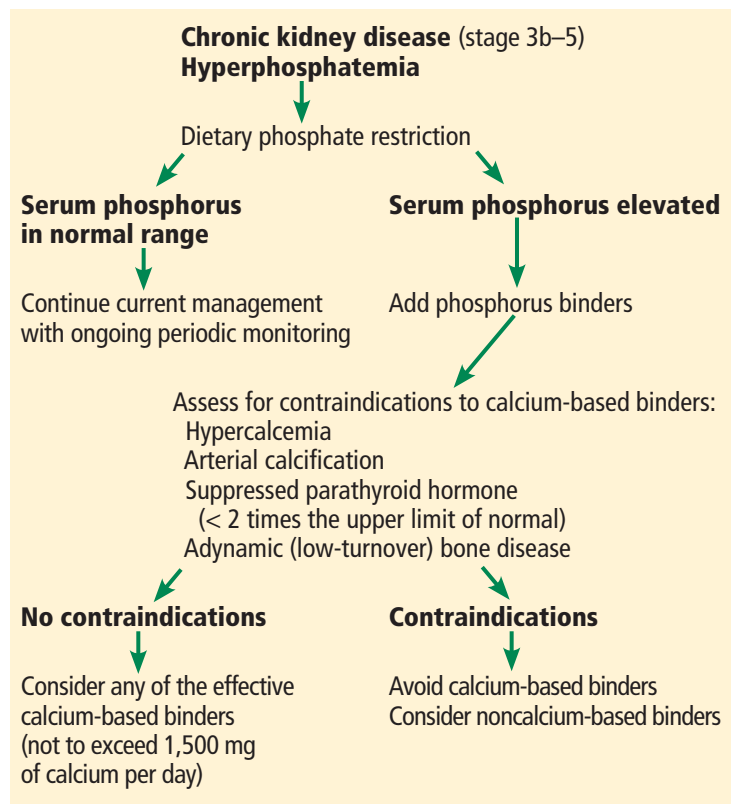
The phosphate content of medications should be considered when prescribing. For example, Reno Caps (Nnodum Pharmaceuticals, Cincinnati, OH), another vitamin supplement, has only 1.7 mg of phosphorus per tablet and should be considered, especially in patients with poorly controlled serum phosphorus levels. However, the challenge is that medication labels do not provide the phosphorus content.

### Reducing phosphorus absorption

Because so many foods contain phosphorus, dietary efforts alone are often insufficient to control serum phosphorus levels, and most patients require additional strategies, eg, phosphorus binders (Table 1).

Although these agents reduce serum phosphorus and help reduce symptoms, an important quality-of-life measure, it is uncertain whether they improve clinical outcomes.<sup>11</sup> To date, no specific phosphorus binder offers a survival benefit over placebo.<sup>11</sup>

Based on the limited and conflicting evidence, the Kidney Disease: Improving Global Outcomes (KDIGO) guidelines, recently updated, suggest that oral phosphorus binders should be used in patients with hyperphosphatemia to lower serum phosphorus levels toward the normal range.<sup>15</sup> They further recommend not exceeding 1,500 mg of elemental calcium per day if a calcium-based binder is used,



**Figure 1.** A stepwise approach to the management of hyperphosphatemia and selection of phosphorus binder.

and they recommend avoiding calcium-based binders in patients with hypercalcemia, adynamic bone disease, or vascular calcification.

Phosphorus binders may account for up to 50% of the daily pill burden and may contribute to poor medication adherence.<sup>16</sup> Dialysis patients need to take a lot of these drugs: by weight, 5 to 6 pounds per year.

These drugs can bind and interfere with the absorption of other vital medications and so should be taken with meals and separately from other medications.

At present, there is insufficient evidence to recommend one binder over the other, and the selection of phosphorus binder should be individualized for each patient, taking into consideration the stage of chronic kidney disease, degree of hyperphosphatemia, concomitant anemia, presence of vascular calcification, use of other medications, side effects, cost to the individual, and pill burden. A stepwise, opinion-based, clinical approach to the selection of the phosphorus binders in patients with hyperphosphatemia is presented in Figure 1.

## Removing phosphorus

Removal of phosphorus by adequate dialysis or kidney transplant is the final strategy.

## New agents under study

To improve phosphorus control, other agents that inhibit absorption of phosphate are being investigated.

**Nicotinamide** reduces expression of the sodium-phosphorus cotransporter NTP2b. Its use in combination with a low-phosphorus

diet and phosphorus binders may maximize reductions in phosphorus absorption and is being studied in the CKD Optimal Management With Binders and Nicotinamide (COMBINE) study.

**Tenapanor**, an inhibitor of the sodium-hydrogen transporter NHE3, has been shown in animal studies to increase fecal phosphate excretion and decrease urinary phosphate excretion<sup>17</sup> but requires further evaluation. ■

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